



UC BERKELEY MICROFABRICATION LAB INTERNSHIP
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Comparison of Thin-Film Aluminum Deposition



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OBJECTIVE

To compare thin-film aluminum deposition using:

- 1) Resistivity
- 2) Uniformity
- 3) Deposition Rate



OUTLINE OF PROJECT

- **Cleaning Wafers & Oxide Growth**
- **Sputtering**
 - Novellus m2i Sputtering System
 - Edwards Auto 306DC and RF Sputter Coater
 - CPA Sputtering System
- **Evaporating**
 - NRC Evaporator
 - Veeco 401 Vacuum System
 - Edwards eb3 Electron Beam Evaporator
- **Photolithography & Etching**



MEASUREMENT TOOLS & METHOD

NANO SPEC

- thickness of oxide
- reflectance



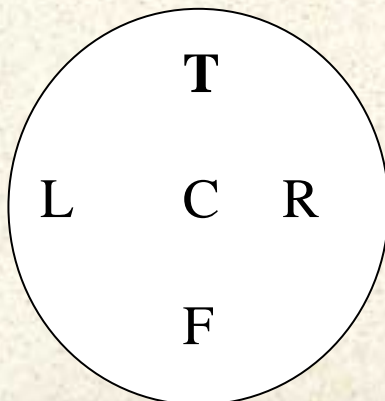
ASIQ

- step measurement



4 POINT PROBE

- resistivity



- systematic measurement of top, center, flat, left, right



CLEANING WAFERS & OXIDE GROWTH

Sink 6

- Piranha (Sulfuric Acid) with H_2O_2
 - strong oxidizer- removes organic contamination
- Creates some SiO_2

- HF- dissolves SiO_2
 - remove metallic contamination

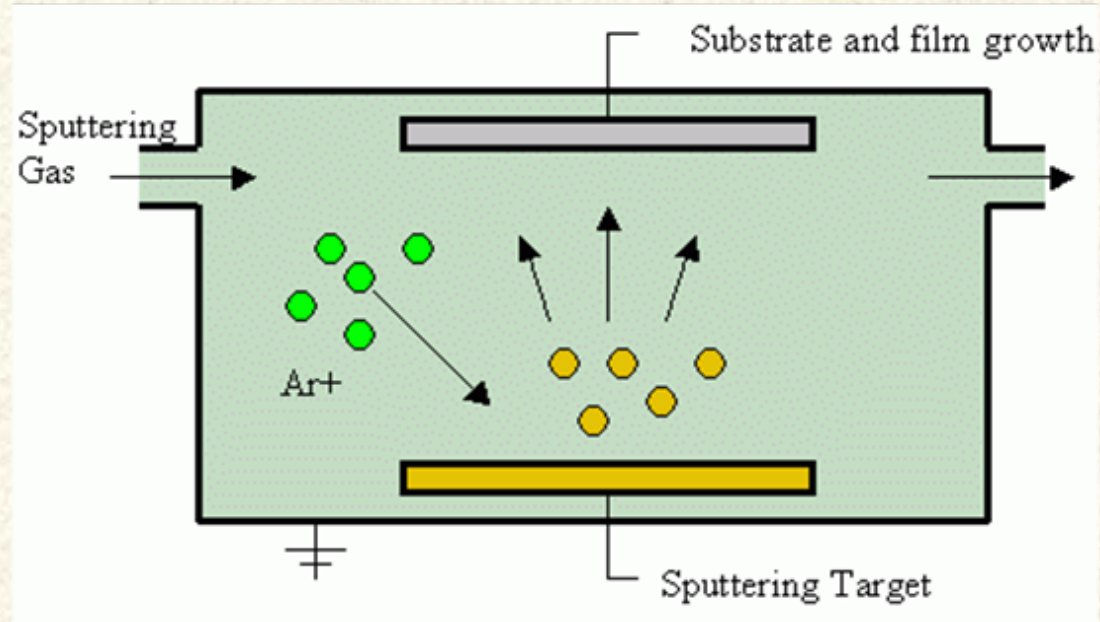
Tystar 3

- Recipe: 3WETOX
- Temperature: $1000^\circ C$
- Time: 10 min 30 sec

- Steam to oxidize Si \rightarrow faster than when dry



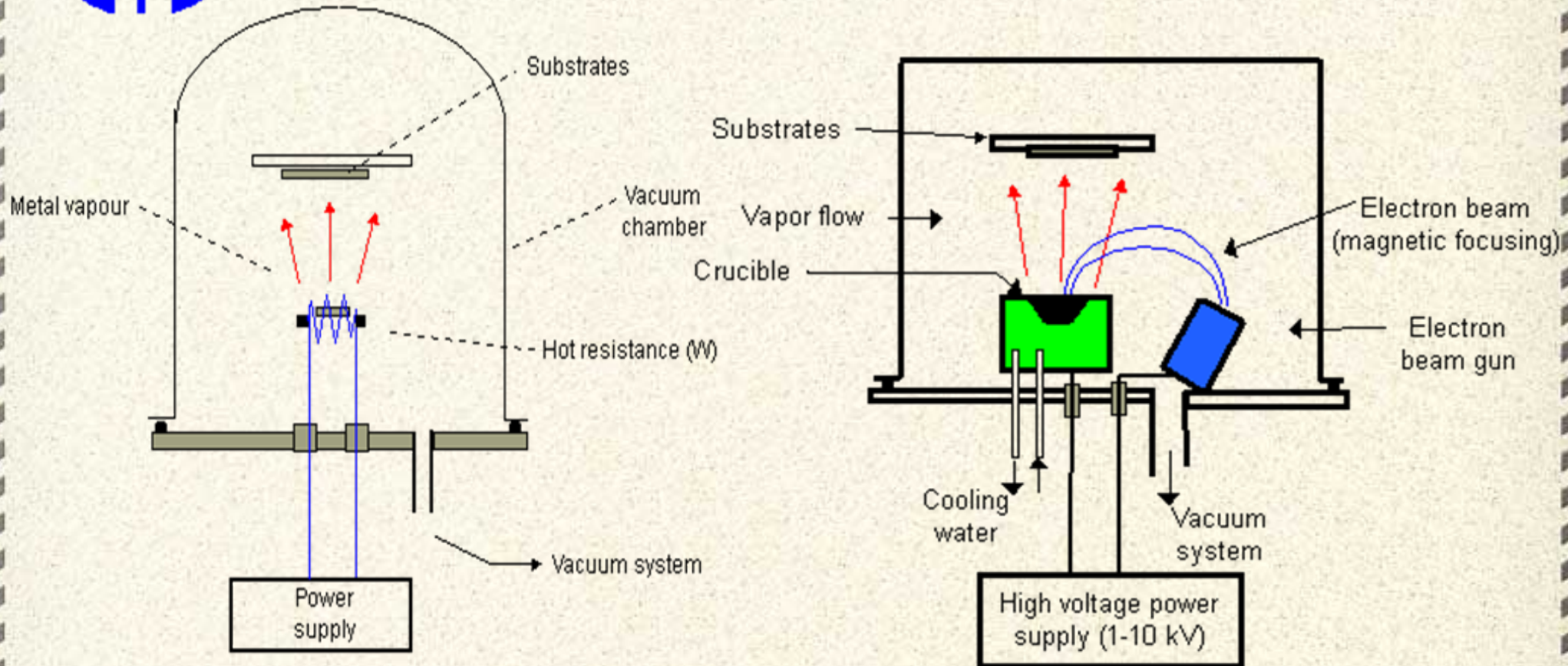
SPUTTERING



Plasma (Argon gas) is ejected into the sputtering target, which releases clusters of aluminum particles onto the substrate (wafer).



EVAPORATING



Metal is heated through a filament, crucible, or metal plate. The evaporated metal is, then, condensed onto the substrate or wafer.

For an electron beam evaporator, an electron beam bombards the metal which evaporates onto the substrate or wafer.



COMPARISON OF SPUTTERING AND EVAPORATING

	Advantages	Disadvantages
Evaporating	<ul style="list-style-type: none">• consumes an efficient amount of aluminum• more cost efficient (little Al per use)• versatile in ability to change metals	<ul style="list-style-type: none">• time consuming
Sputtering	<ul style="list-style-type: none">• quick aluminum deposition	<ul style="list-style-type: none">• large machine (space is necessary)• expensive aluminum sheet for large target• difficult to change metal



DATA ANALYSIS

Wafer	Non-Uniformity (%)	Bulk Resistivity (Ω -cm)	Overall Time (Hr)	Deposition Rate ($\text{\AA}/\text{min}$)
Novellus (25)	13.45	0.0414392	1/6	279.6
Novellus (26)	11.06	0.02634984	1/6	550
Edwards (13)	10.84	0.1126076	4	53.44
CPA (2)	1.1	0.03399536	4	N/A
NRC (5)	9.48	0.7033494	2	242.47
V401 (12)	12.39	0.184477	2	251.8
Edwards eb3 (31)	21.95	0.06602736	4	150

- Non-Uniformity= (max-min)/avg
- Bulk Resistivity= resistivity*thickness



DATA ANALYSIS

Wafer	Pre-thickness of Oxide	Post-thickness of Oxide	Reflectance at 640 nm (% relative to Si)
Novellus (25)	1084.4	1029.2	244.8
Novellus (26)	1218.4	1041.8	226
Edwards (13)	984.8	1007.6	217.2
CPA (2)	999.4	1432.8	234.4
NRC (5)	1001.4	1002.4	160.4
V401 (12)	1004.8	1004.8	212.2
Edwards eb3 (31)	1237.4	1193	196



OBSTACLES

- When using NRC(1) and V401(2),

- 1) a shadow formed, not allowing a even coat of aluminum

- 2) the aluminum became tinted with a golden color

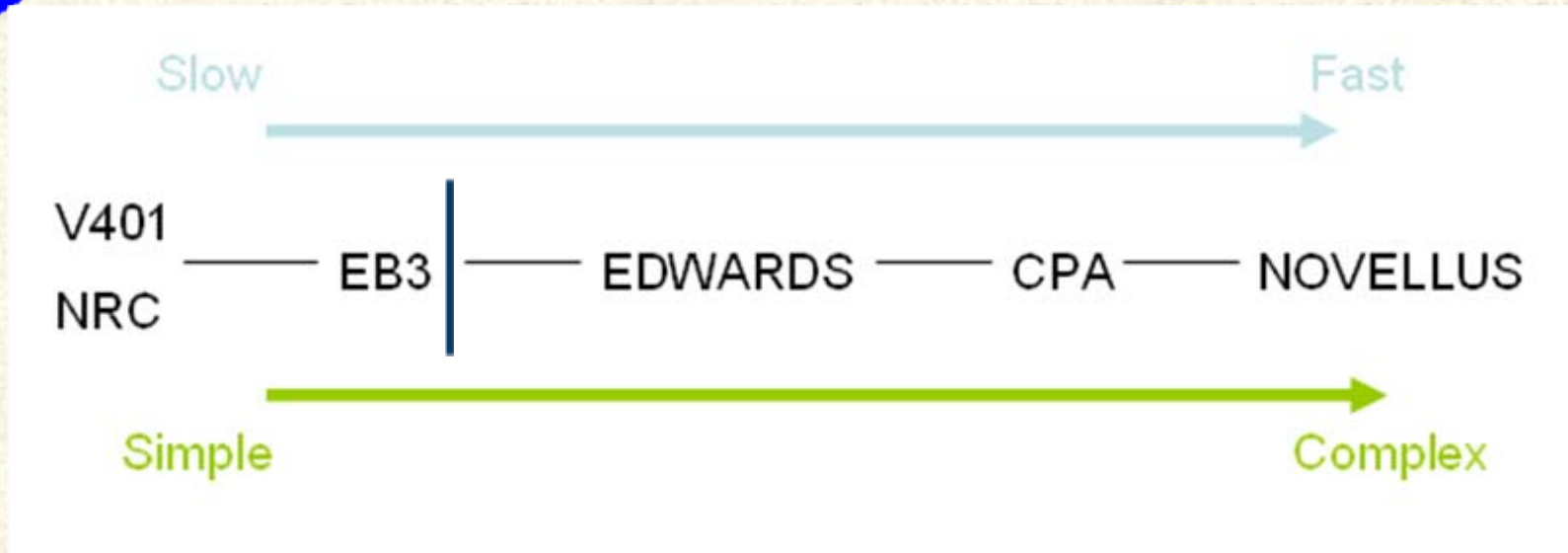
Solution: Re-do 2 wafers for each machine

- When developing, some of the photo-resist did not develop properly, creating an inability for the CPA wafer to etch.

Solution: Develop the wafer for a longer duration of time by developing twice



CONCLUSION



- In a lab setting, evaporating is advantageous because it is cost efficient. Although evaporating is time consuming, labs are not mass producing, therefore, mitigating the problem of time.
- Novellus outperforms in deposition rate, time, resistivity, and reflectance.
- CPA produces the most uniform layer of aluminum.



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